

GUJARAT TECHNOLOGICAL UNIVERSITY
BE – SEMESTER–VIII • Remedial EXAMINATION – WINTER 2013

Subject Code: 180505**Date: 12/09/2013****Subject Name: Multi Component Distillation****Time: 03:00 pm – 05:30 pm****Total Marks: 70****Instructions:**

1. Attempt all questions.
2. Make suitable assumptions wherever necessary.
3. Figures to the right indicate full marks.

- Q.1 (a)** In the production of chloromethanes, a saturated liquid mixture consisting of 60% methyl chloride(CH_3Cl), 28% methylene chloride(CH_2Cl_2), 9% chloroform(CHCl_3) and 3% carbon tetra chloride(CCl_4)(by mass), is sent to distillation for separation of pure products. In first distillation column 99.9% (by mass) pure methyl chloride is separated as top product and 738 kmol/h pure methyl chloride is present in distillate. Feed is saturated liquid mixture and flow rate is 1000 kmol/h. **14**

Component	Average Relative Volatility
Methyl chloride	6.224
Methylene chloride	1
Chloroform	0.4954
Carbon tetra chloride	0.3126

Determine the following of distillation column:

- (a) Do the complete material balance.
 - (b) Minimum reflux ratio by Underwood's method
 - (c) If $R = 3R_m$, calculate the number of theoretical stages required for desired separation.
- Q.2 (a)** Discuss various factors that must be considered for the selection of tray type in tray tower. **07**
- (b)** (i) What is MESH equation? In multicomponent distillation, for C components and N number of equilibrium stages how many MESH equations can be written? **04**
- (ii) What are the disadvantages of vacuum distillation? **03**
- OR**
- (b)** Discuss in detail about selection of operating pressure for distillation column. **07**
- Q.3 (a)** List out the variables that are to be specified as input data for Thiele-Geddes method. Explain the stepwise procedure of Thiele Geddes method for stripping section of Multicomponent distillation. **10**
- (b)** How will you identify feed tray location in Lewis –Matheson method of Multicomponent distillation? **04**
- OR**
- Q.3 (a)** Explain the method of determining tower diameter in sieve tray tower. **07**
- (b)** Discuss heuristics for determining favorable sequences in distillation column. **07**

- Q.4 (a)** A saturated liquid, consisting of phenol and cresols with some xylenols, is fractionated to give a top product of 95.3 mole % phenol. Metacresol is heavy key and phenol is light key component. Total condenser is used. The compositions of the top product and of the phenols in the bottoms are given. **14**

Component	AverageRelative Volatility	Feed, mole %	Top product, mole %	Bottom product, mole %
Phenol	1.98	35	95.3	5.24
o-Cresol	1.59	15	4.55	?
m-Cresol	1.00	30	0.15	?
Xylenols	0.59	20	-	?

- (a) Compute the material balance over the still for a feed rate of 100 kmol/h.
 (b) If $R=3R_m$, calculate the composition of vapour entering to the top most tray by Lewis –Matheson method. Assume $R_m= 1.67$.

OR

- Q.4 (a)** Explain the step wise procedure for the process design of multicomponent batch distillation with rectification. **12**
(b) Write down Hegstebeck and Geddes equation for approximate distribution of non key components in top product and bottom product. **02**

- Q.5 (a)** List out the various design options for energy conservation in distillation column. Also discuss the energy saving in distillation column by direct vapour compressor. **09**
(b) Discuss selection criteria of solvent for successful extractive distillation sequence. **05**

OR

- Q.5 (a)** Explain concept and working principle of azeotropic distillation with industrial example. **09**
(b) Explain the concept of heat integration in distillation column. **05**
