

Seat No.: _____

Enrolment No. _____

GUJARAT TECHNOLOGICAL UNIVERSITY
B. Pharm. – SEMESTER – I • EXAMINATION – SUMMER 2013

Subject Code: 2210002

Date: 13-05-2013

Subject Name: Pharm Chemistry-I (Inorganic chemistry)

Time: 02.30 pm - 05.30 pm

Total Marks: 80

Instructions:

- 1. Attempt any five questions.**
- 2. Make suitable assumptions wherever necessary.**
- 3. Figures to the right indicate full marks.**

- Q.1** (a) Discuss the source of impurities in pharmaceutical substances. **06**
(b) Define Limit test. Explain the limit test of arsenic. **05**
(c) Write a note on chelating agents used in therapy. **05**
- Q.2** (a) Define buffers. Give mechanism of buffer action and importance of buffer solution in pharmacy. **06**
(b) Explain Bronsted acid base theory. What are conjugate pairs of acid and base? **05**
(c) Give methods of preparation, properties and uses of hydrochloric acid. **05**
- Q.3** (a) Classify gastro-intestinal agents with examples. **06**
(b) Define antacids and explains one of the aluminum compounds as antacids. **05**
(c) Define antioxidants. List the compounds of sodium as antioxidants and give method of preparation of any one. **05**
- Q.4** (a) Define following terminology:- (Any three) **06**
I) Antiseptic II) Filter-Aid III) Adsorbent IV) Preservative
(b) What are topical agents? Classify them with suitable examples. **05**
(c) Enlist the official preparations of iodine and give mechanism of action. **05**
- Q.5** (a) What are anticaries agents? Discuss the role of fluoride. **06**
(b) Give the storage condition of following gases. **05**
I) Carbon dioxide II) Nitrous oxide III) Oxygen
(c) Define expectorants. Discuss the role of ammonium compounds as respiratory stimulants. **05**
- Q. 6** (a) Enlist major intra and extra cellular electrolytes. Discuss the physiological importance of sodium ion. **06**
(b) Write a short note on ORS. **05**
(c) What is haematinics? Give preparation, properties and uses of ferrous gluconate. **05**
- Q. 7** (a) Write a note on geiger muller counter. **06**
(b) Define antidotes. Discuss mechanism of action of antidote poisoning. **05**
(c) Discuss biological importance of calcium ions and their deficiency symptoms. **05**
